

INFLUENCE OF THE CALCINATION ATMOSPHERE ON COPPER-ALUMINUM CATALYSTS FOR THE SELECTIVE DEHYDRATION OF GLYCEROL

Alejandro Lete*, Francisco Lacleta, Lucía García, Joaquín Ruiz, Jesús Arauzo

*alete@unizar.es

Thermochemical Processes Group (GPT), Instituto de Investigación en Ingeniería de Aragón (I3A), Universidad de Zaragoza, Mariano Esquillor S/N, 50018, Zaragoza, Spain

CONTEXT



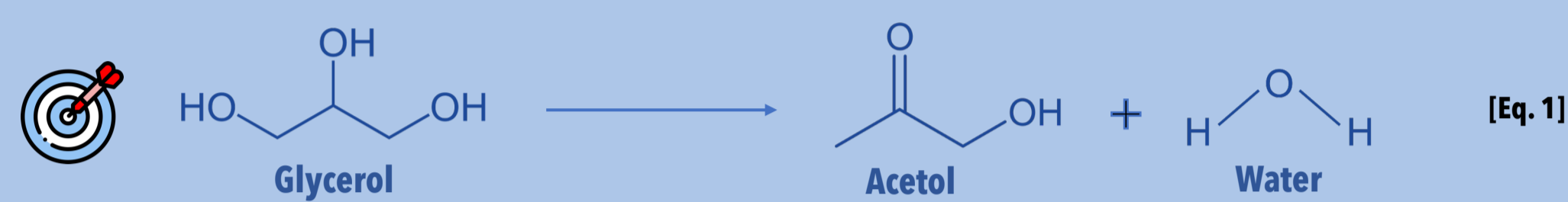
The aviation sector consumes **10.4% of the world's petroleum** [1]. To achieve the goal of zero net carbon emissions by **2050**, the development of sustainable aviation biofuels (SAF) is necessary.



Sugar to Jet Fuels (STJ) is a promising new route that uses biomass-derived furfural as feedstock for SAF production. However, acetol is used in an aldol condensation step and is made from fossil resources [2].



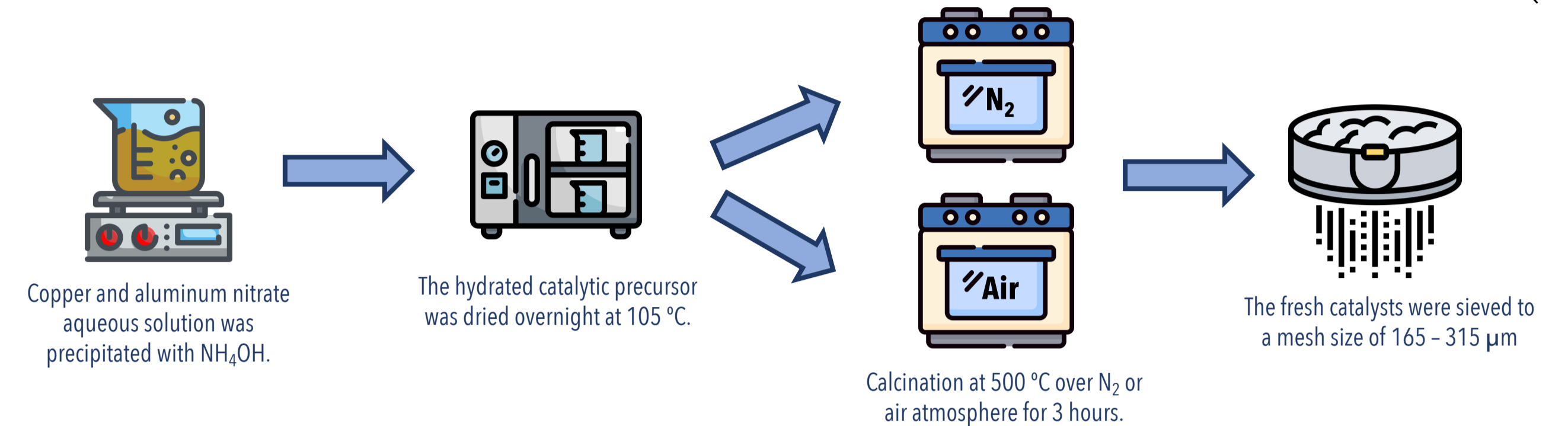
Glycerol, generated as a by-product in the biodiesel manufactured by transesterification of vegetable oils and animal fats, is proposed as feedstock for acetol production by dehydration [Eq. 1].



This work studied the influence of the **calcination atmosphere** (nitrogen or air) in CuAl catalysts for the catalytic production of **acetol** from glycerol, a renewable resource.

METHODOLOGY

The CuAl catalysts were synthesized by the **coprecipitation** method with 28% molar of Cu defined as $\left(\frac{\text{Cu}}{\text{Cu+Al}}\right)$.



Operating conditions

- Atmospheric pressure, gas phase experimental plant (Fig. 1)
- Temperature: 250 °C
- W/m ratio of 30 g_{Cat} min g_{Glycerol}⁻¹
- 10 wt.% glycerol aqueous solution

CATALYTIC ACTIVITY

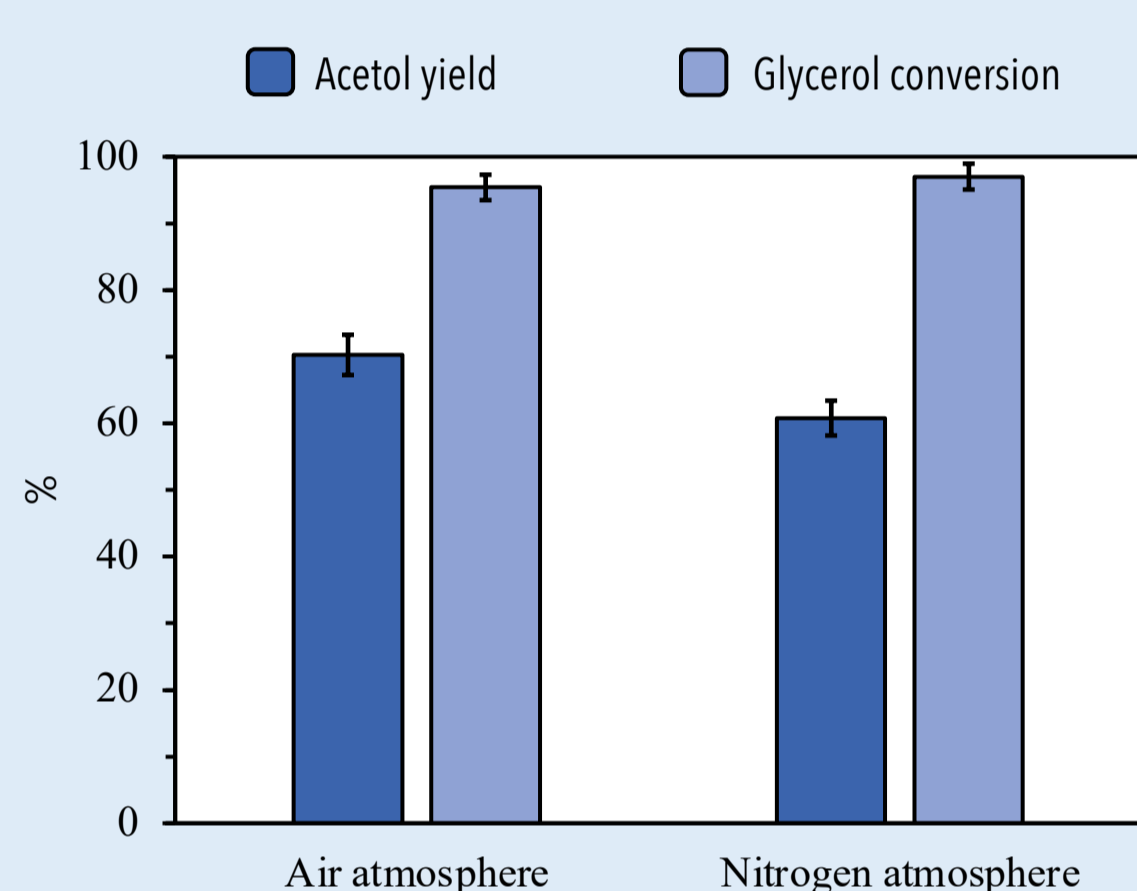


Figure 2. Catalytic performance of CuAl catalysts.

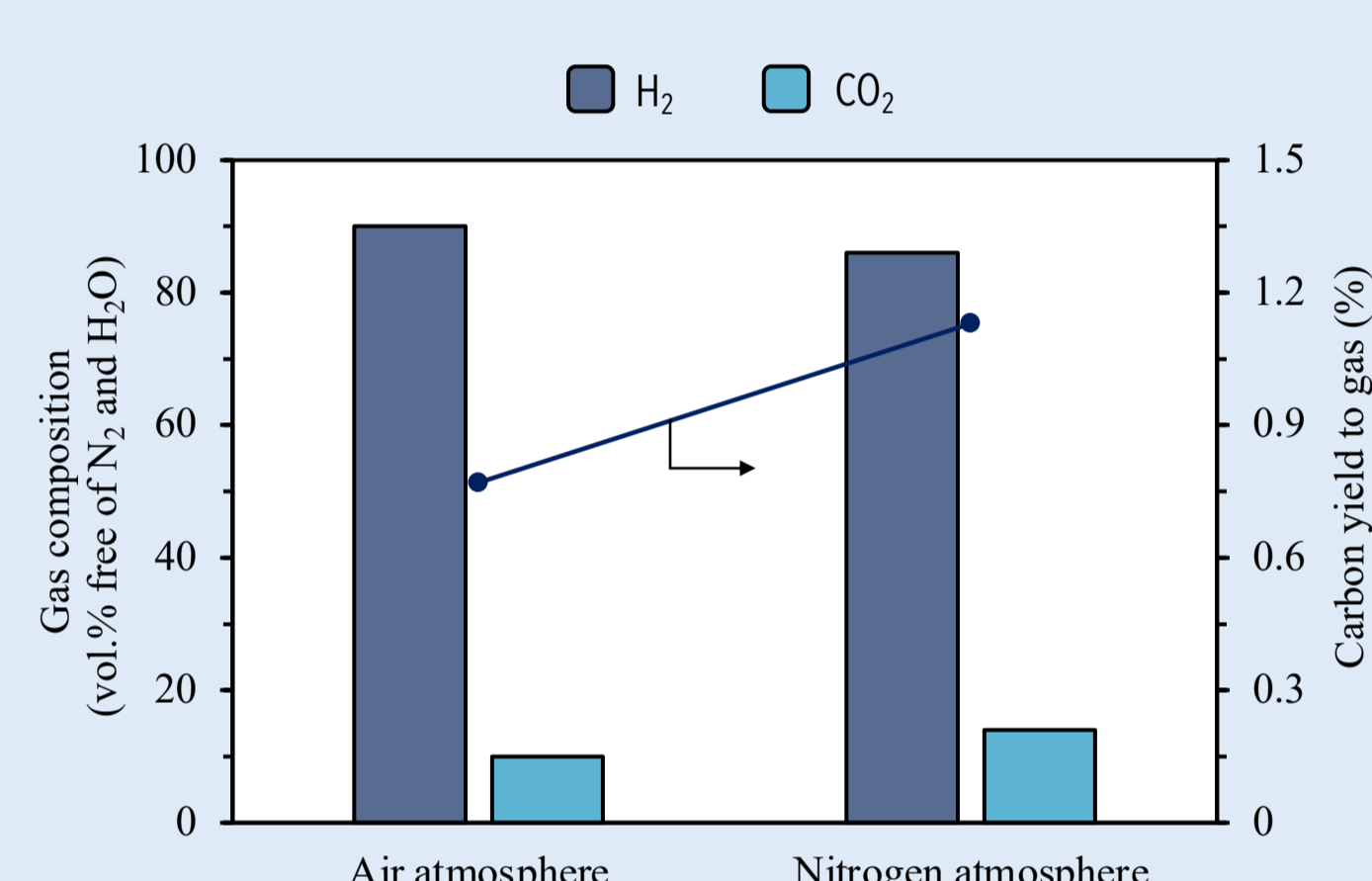


Figure 3. Gas composition and yield with the CuAl catalysts.

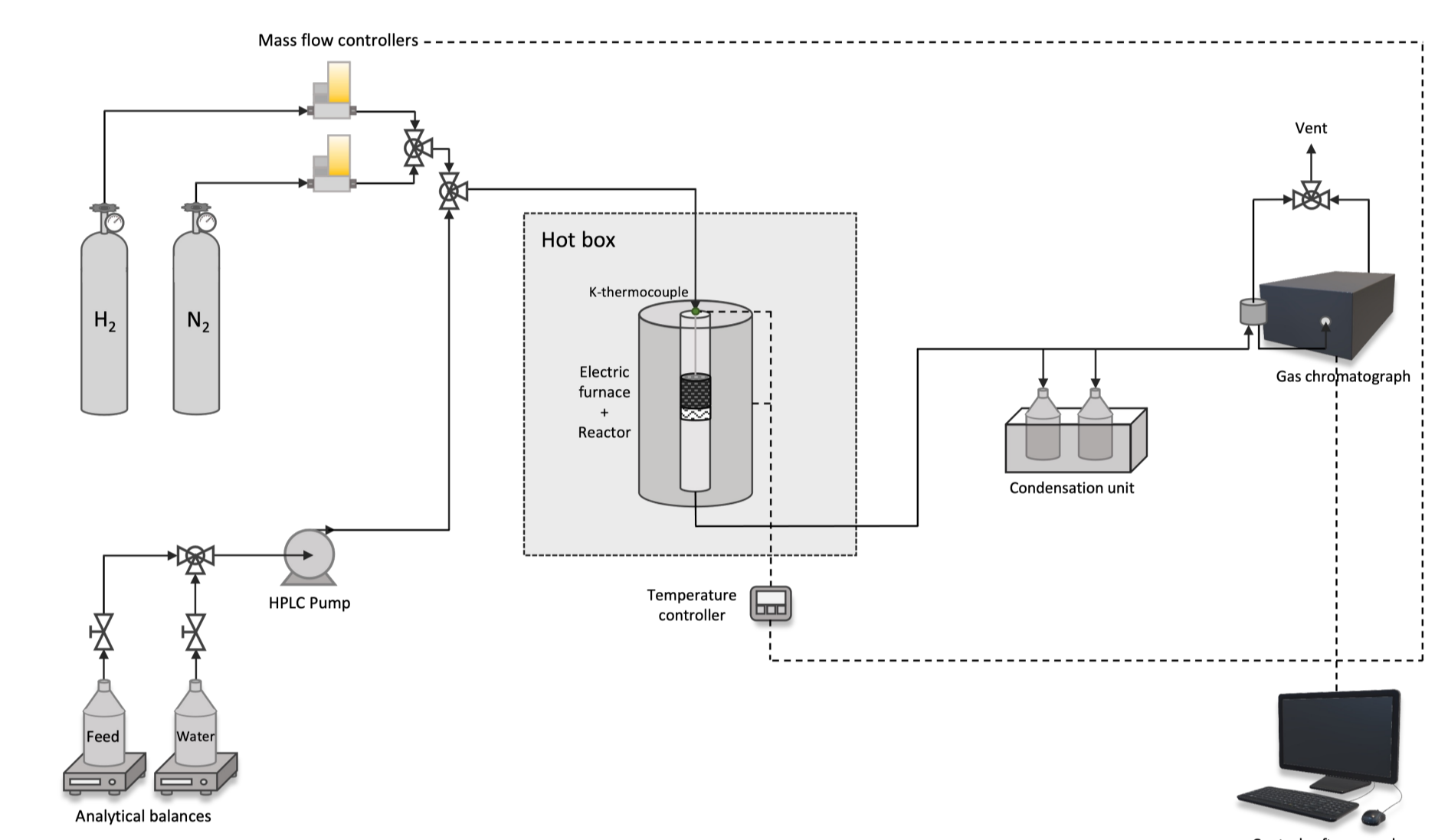


Figure 1. Laboratory scale experimental plant [3].

CHARACTERIZATION

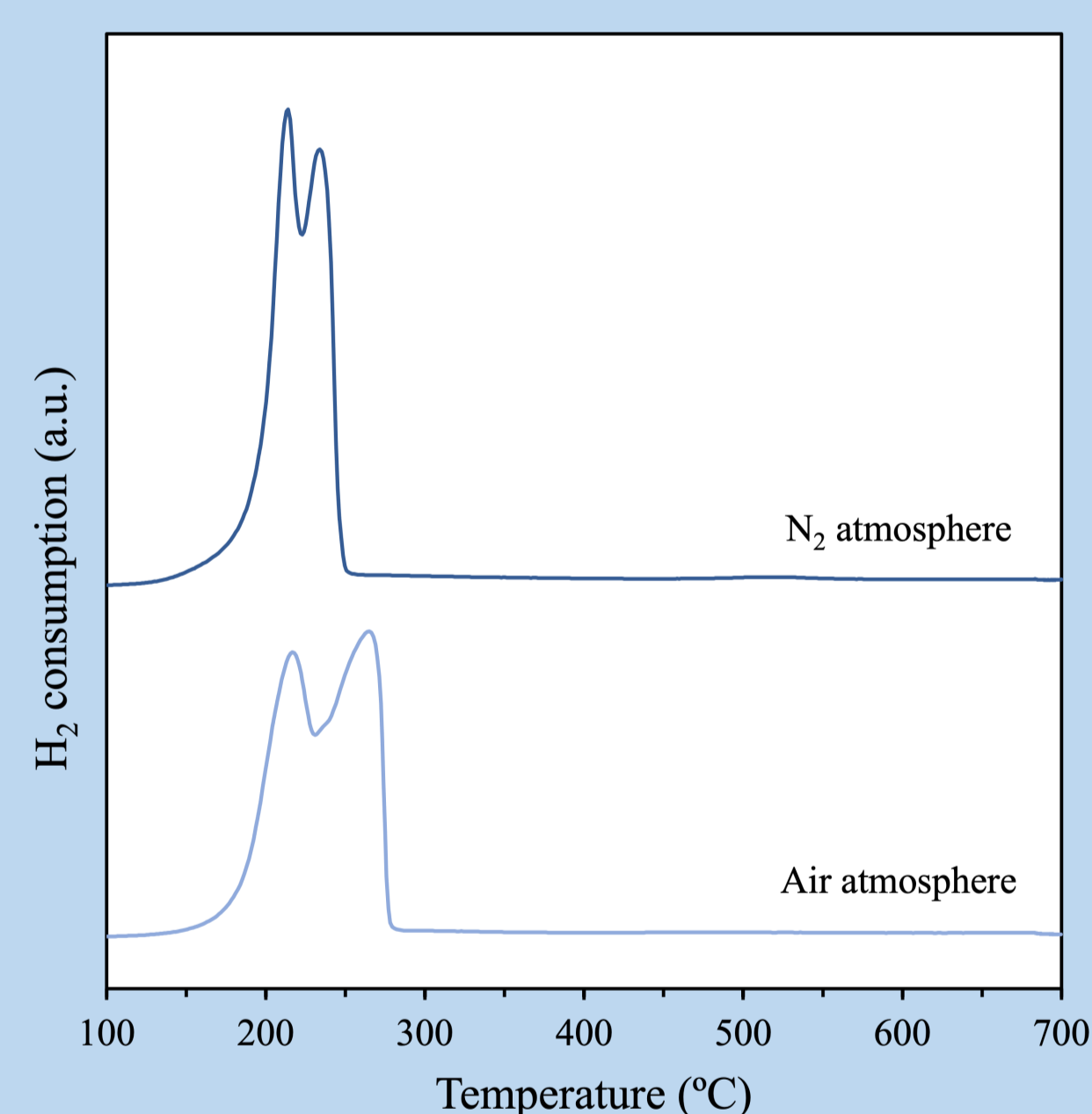


Figure 1. H₂-TPR pattern of fresh catalysts.

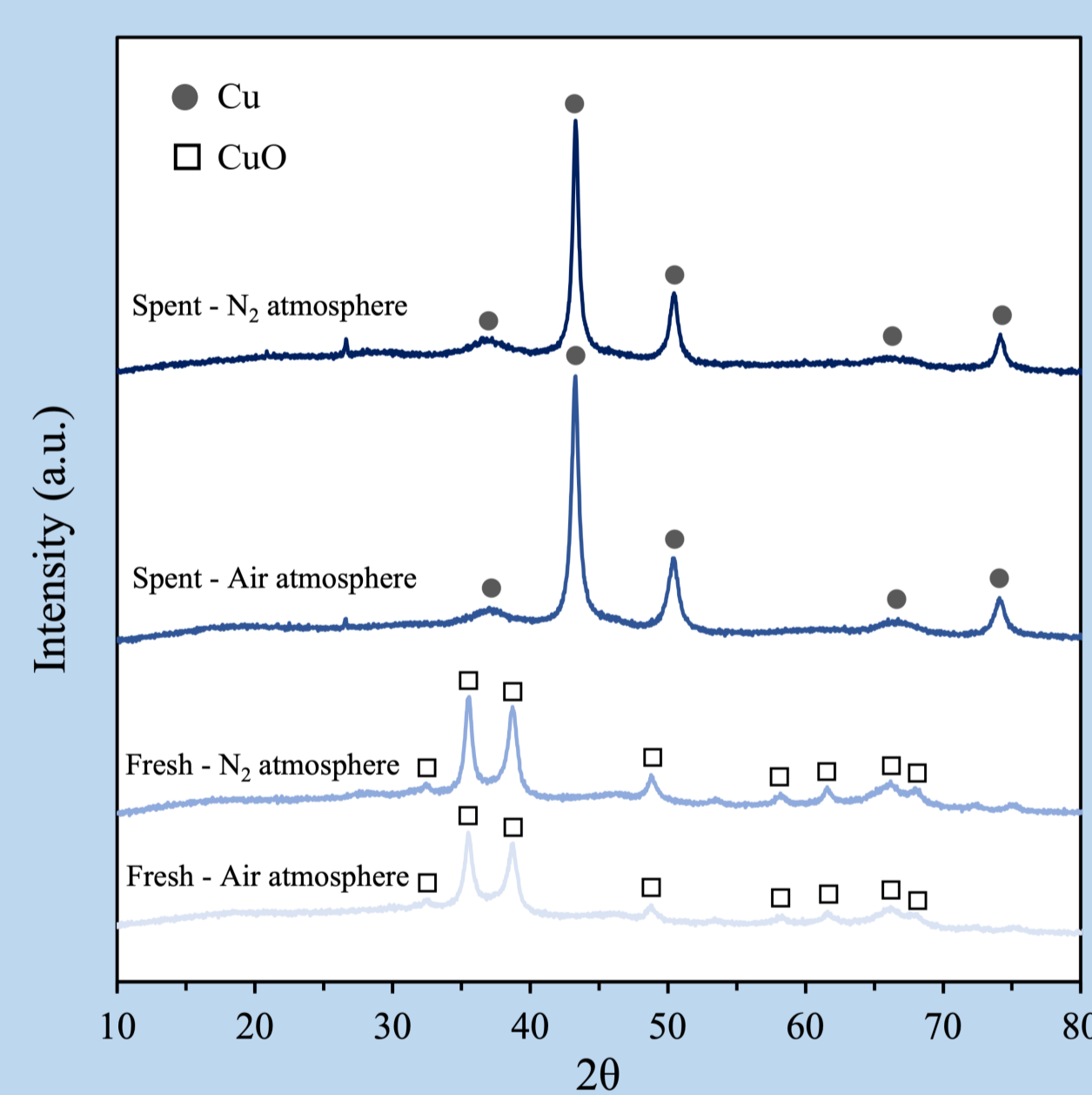


Figure 2. XRD profiles of fresh and spent catalysts.

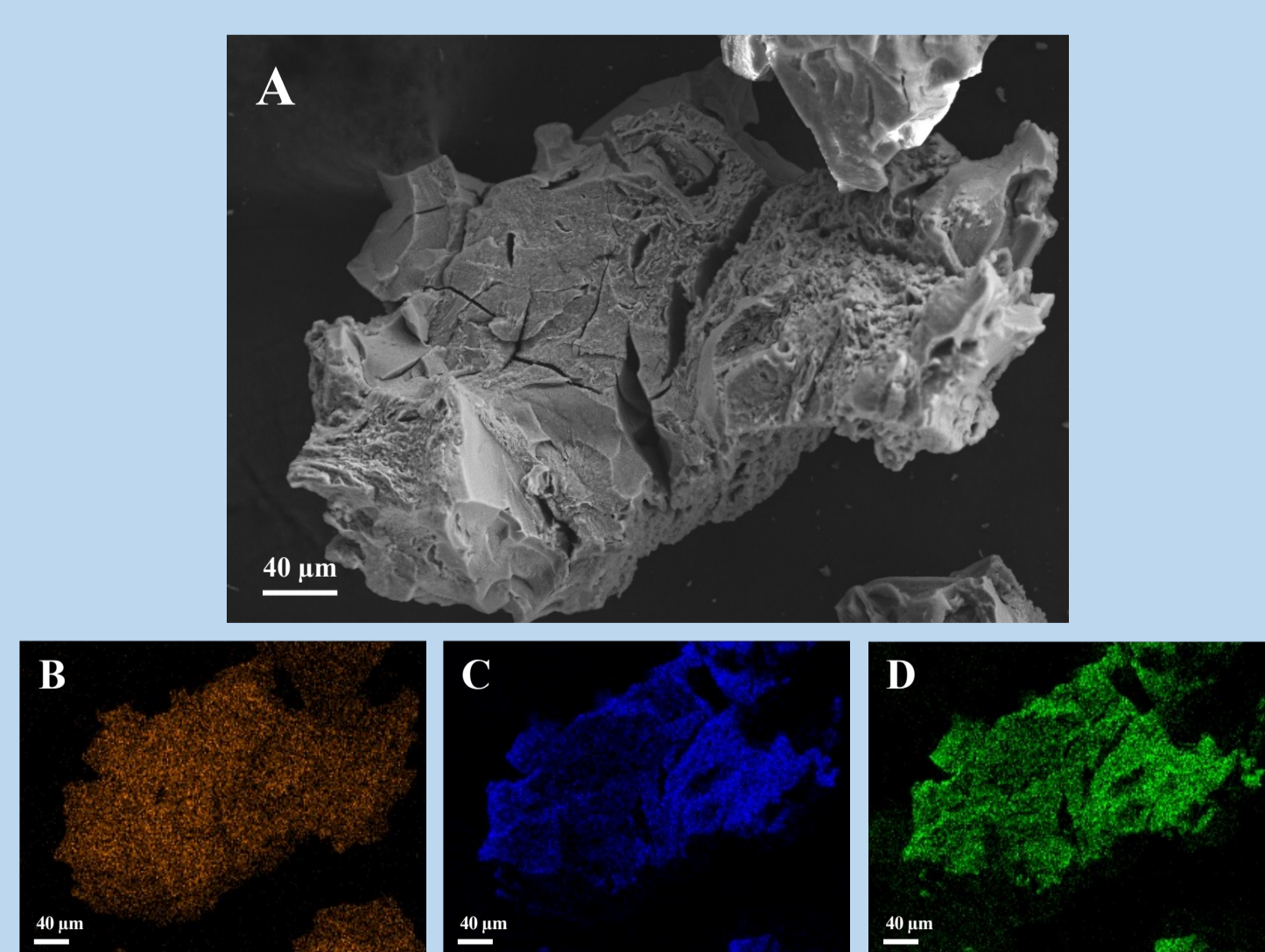


Figure 3. FESEM of fresh catalyst calcined in air atmosphere (A) and EDS mapping of copper (B), aluminum (C), and oxygen (D).

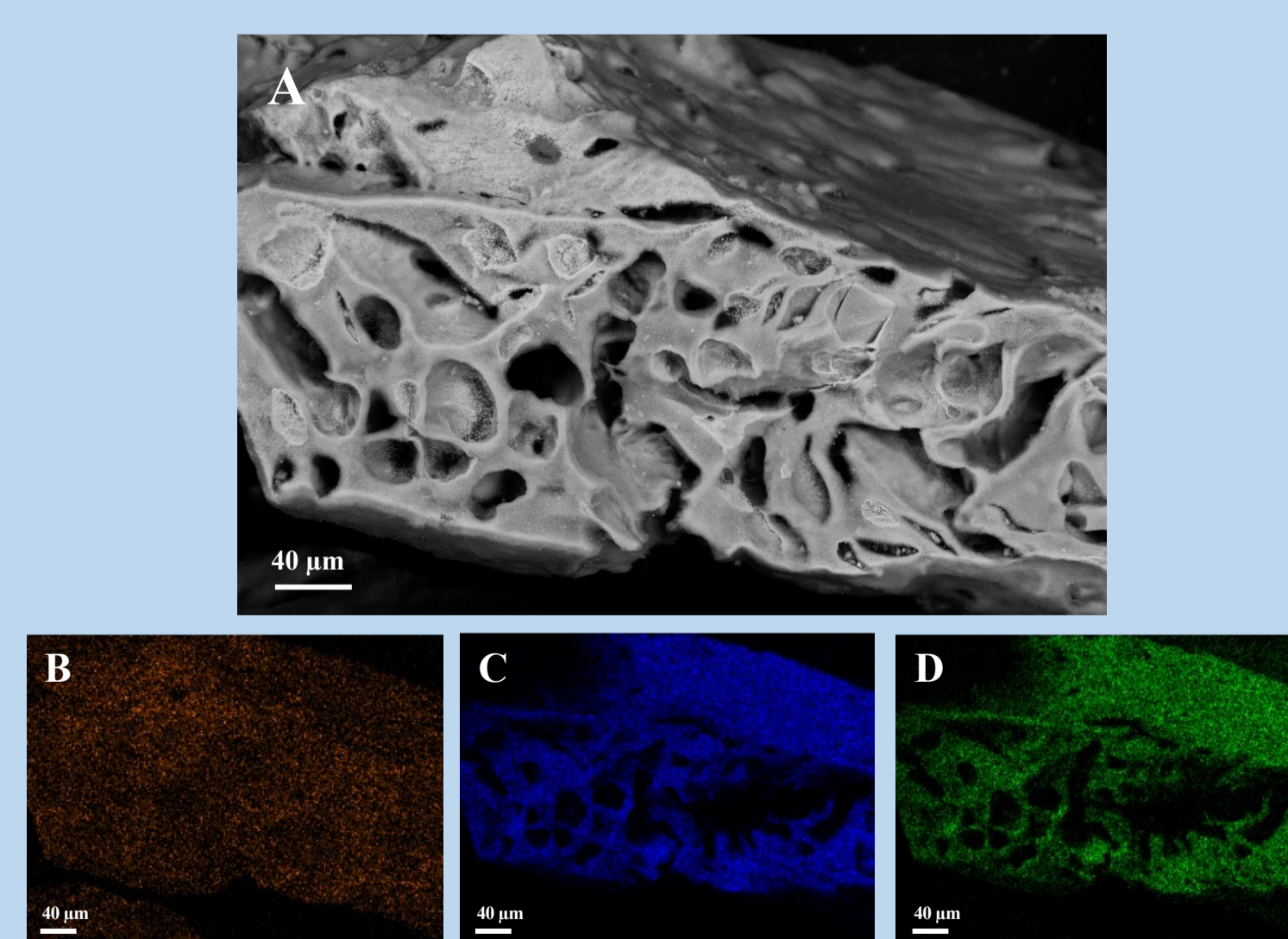


Figure 4. FESEM of fresh catalyst calcined in nitrogen atmosphere (A) and EDS mapping of copper (B), aluminum (C), and oxygen (D).

Table 1. Textural properties of fresh and spent catalysts.

Catalyst	S _{BET} (m ² g ⁻¹)	V _P (cm ³ g ⁻¹)	d _P (nm)	D _{XRD} ^a (nm)
Fresh – Air	224.5	0.40	6.73	13.4
Fresh – Nitrogen	226.7	0.48	6.26	14.5
Spent - Air	219.5	0.36	5.79	14.7
Spent – Nitrogen	165.8	0.34	6.76	18.5

^a Particle CuO and Cu size for the fresh and spent catalysts, respectively. Calculated from XRD patterns by Scherrer equation.

CONCLUSIONS

- The calcination atmosphere influenced the physicochemical properties of the CuAl catalysts.
- Calcination in N₂ atmosphere lowered the reduction temperature of Cu but reduced its dispersion.
- The air atmosphere generated better textural properties after the reaction.
- The catalyst calcined in air atmosphere improved the acetol yield to 71.0 %.

REFERENCES

- IEA. *Tracking Clean Energy Progress 2023*. IEA, Paris, International Energy Agency, 2023.
- West, R., Liu, Z., Peter, M., Gartner, C. and Dumesic, J. *Carbon-Carbon bond formation for biomass-derived furfurals and ketones by aldol condensation in a biphasic system*. Journal of Molecular A-Chemical. 2008, 296, 18-27.
- Lete, A., Raso, R., García, L., Ruiz, J. and Arauzo, J. *Synthesis of ketones from glycerol and 1,2-propanediol using copper and nickel catalysts: Unraveling the impact of reaction phase and active metal*. Fuel. 2024, 371(A), 132001.