

EXPLORING CLEANER ROUTES OF ALUMINA PRODUCTION: SIMULATION OF PEDERSEN PROCESS AND CO₂ CAPTURE INTEGRATION VIA CALCIUM LOOPING



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XIII JORNADA DE JÓVENES INVESTIGADORES/AS DEL I3A
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Background

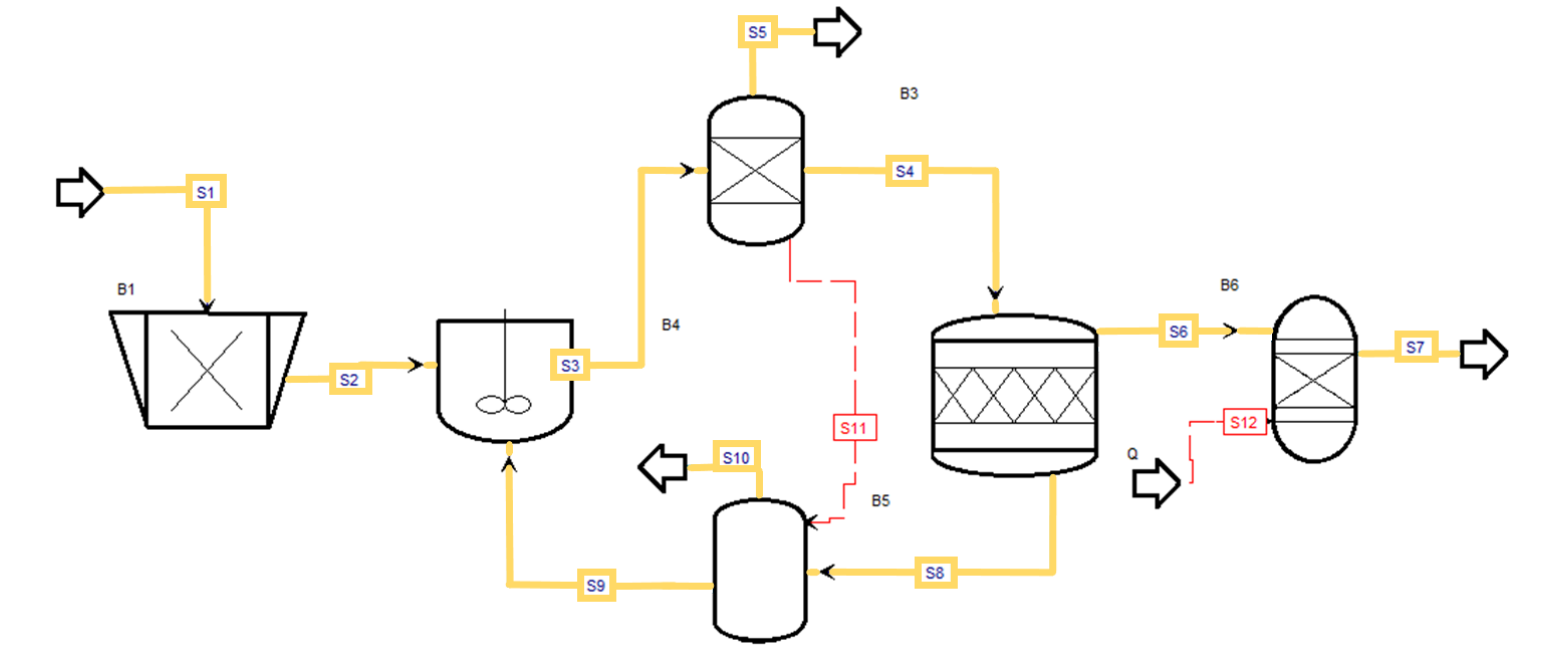
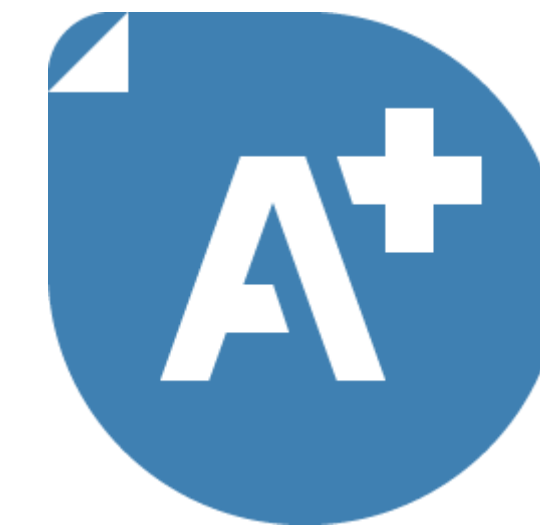
Alumina (Al₂O₃) is a critical raw material for today's society. It is extracted from bauxite ore via the Bayer process. This process is a significant emitter of CO₂ and an important contributor to mineral scarcity because of the generation of big amounts of bauxite residues.

Alternative routes are being explored to produce alumina while avoiding bauxite residues. One of these routes is known as the Pedersen process, which allows the co-production of alumina and pig iron from bauxite.

OBJECTIVES

- Evaluate the energy performance of Pedersen process vs the state of the industry.
- Study the integration of a CO₂ capture plant, exploiting the use of CaCO₃.

Methodology



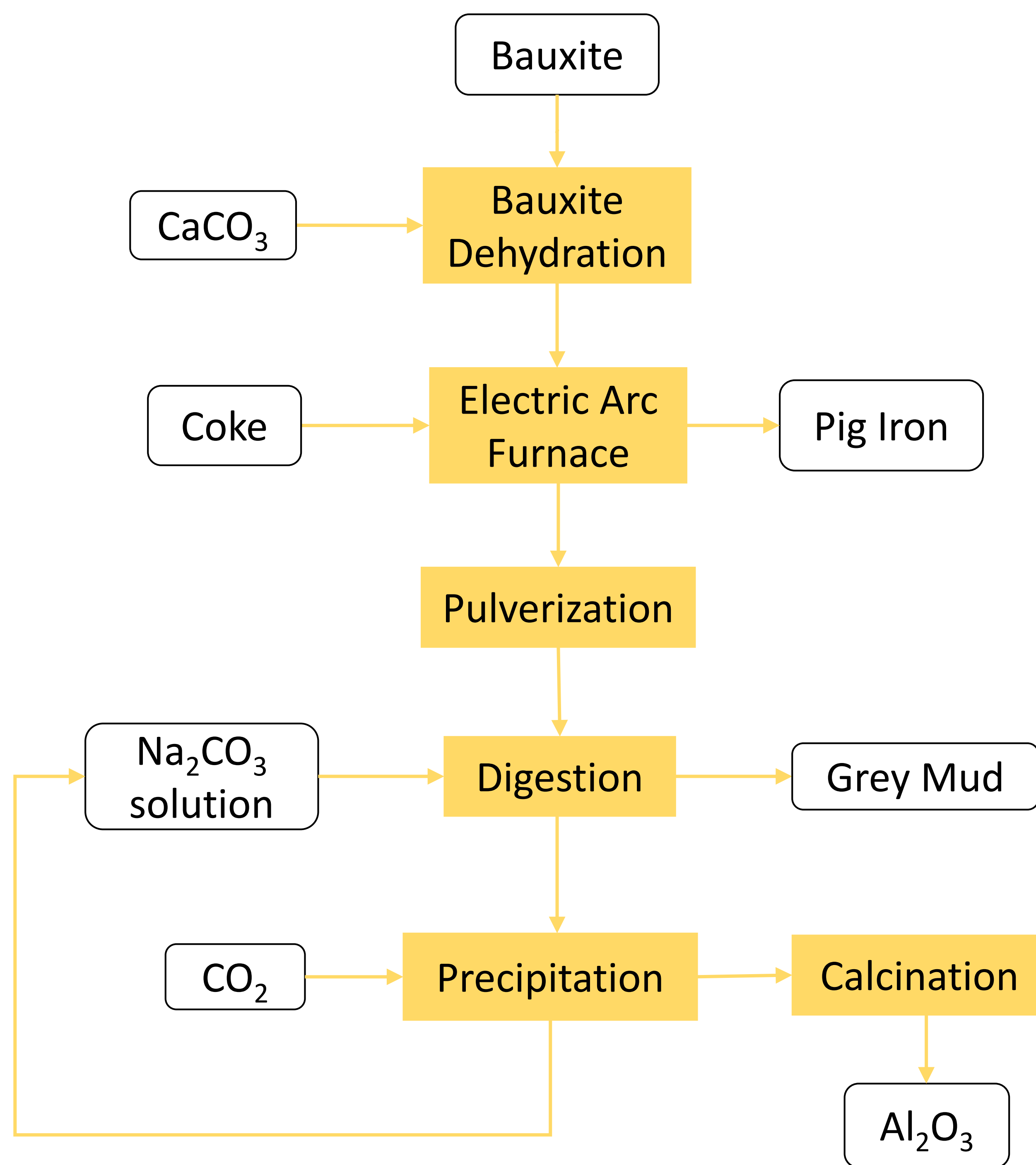
Bauxite Composition¹ (%wt)

Al(OH) ₃	80.1
Fe ₂ O ₃	15.7
SiO ₂	2.7
TiO ₂	1.5

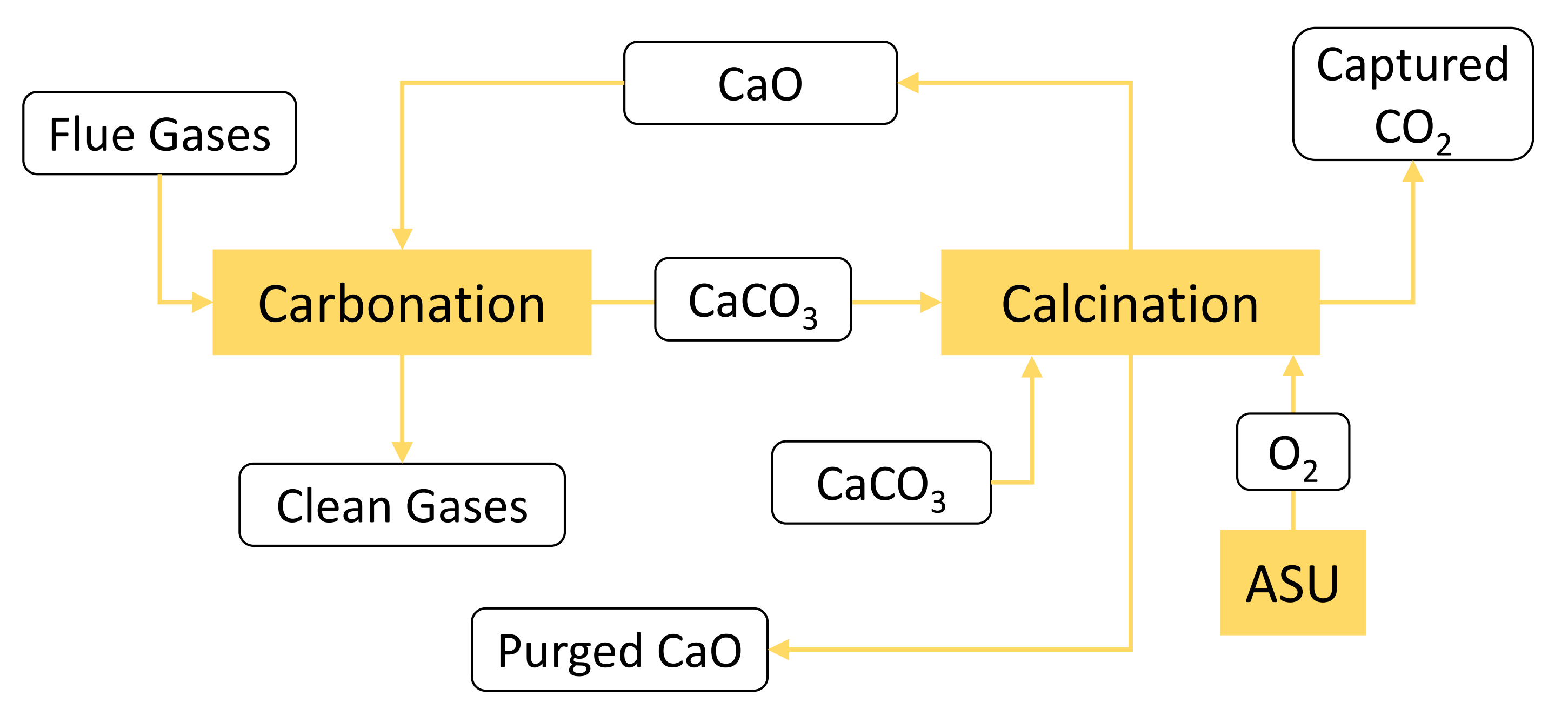
Sorbent Characteristics²

Limestone (CaCO ₃)	
Purge Ratio	2-50%
% CO ₂ Capture	98%

Pedersen Process



Calcium Looping



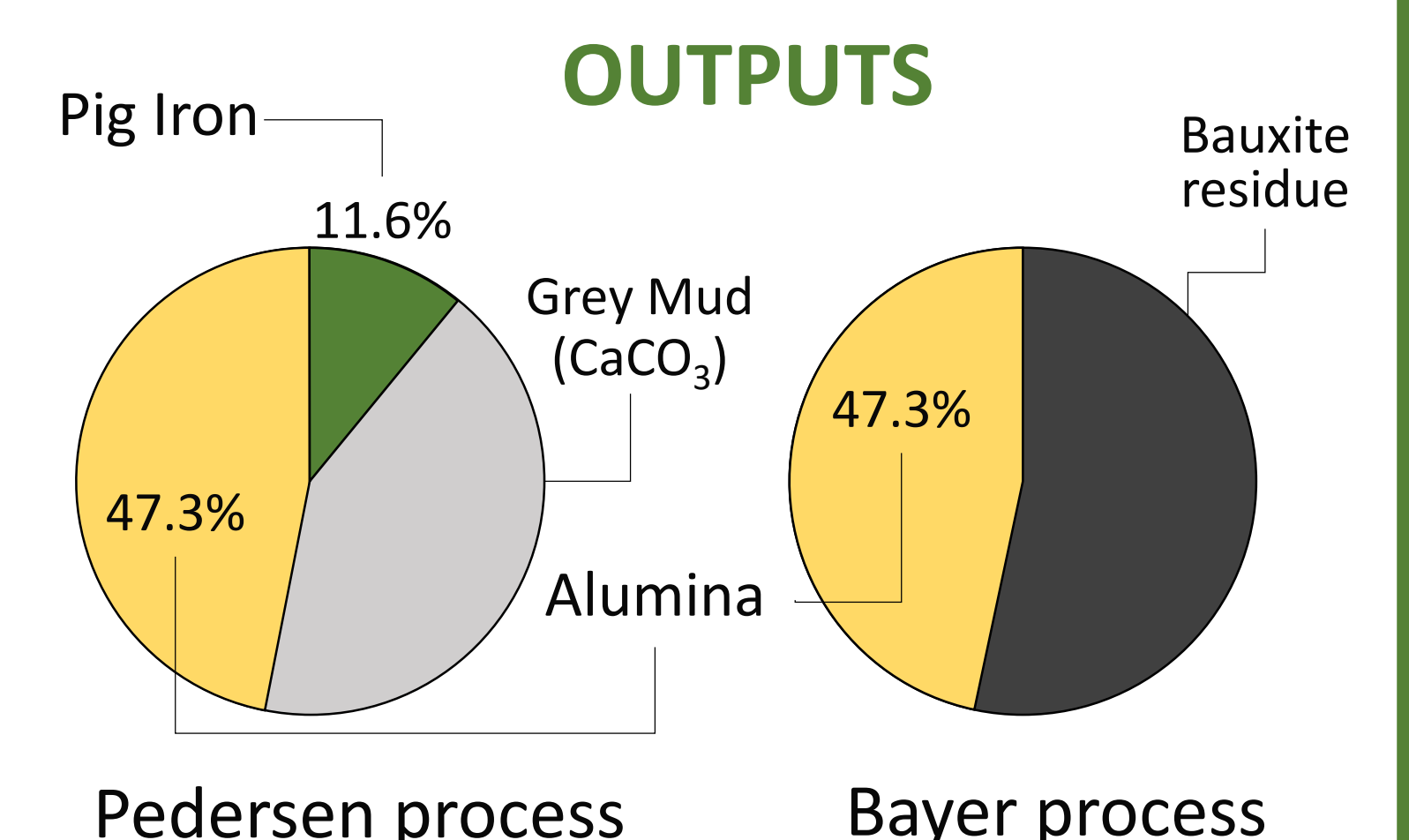
Results

- **0.473 tonnes of Al₂O₃ and 0.116 tonnes of pig iron can be obtained from 1 tonne of bauxite.**
- Energy demand is **13.62 GJ per tonne of Al₂O₃**, significantly higher than average Bayer process (≈10.22 GJ).

Calcium-looping integration	Energy Demand	Energy penalty
Single Plant	13.62 GJ/ton Al ₂ O ₃	-
2.6% Purge ratio	20.29 GJ/ton Al ₂ O ₃	7.94 GJ/ton CO ₂ avoid
33.3% Purge ratio	16.32 GJ/ton Al ₂ O ₃	3.20 GJ/ton CO ₂ avoid
50.0% Purge ratio	15.01 GJ/ton Al ₂ O ₃	1.65 GJ/ton CO ₂ avoid

Conclusions

- Energy consumption of Pedersen process is significantly higher than industrial Bayer process, although **economic assessment should indicate its feasibility** considering pig iron co-production.
- Calcium looping stands out as a favourable strategy to mitigate CO₂ emissions. The grade of **substitution of CaCO₃ by purged CaO** is crucial to decrease the energy penalty of the capture.
- Findings bring remarkable evidence about the possibilities for smart production of two key raw materials **reducing climate change contributions and mineral scarcity.**



REFERENCES

¹ Hudson, L. K., Misra, C., Perrotta, A. J., Wefers, K., Williams, F. S., Aluminum Oxide, 2012. Ullmann's Encyclopedia of Industrial Chemistry, 607-644, doi.org/10.1002/14356007.a01_557

² Pascual, S., Lisbona, P., Bailera, M., Romeo, L. M., Design and operational performance maps of calcium looping thermochemical energy storage for concentrating solar power plants, 2021. Energy 220, doi.org/10.1016/J.ENERGY.2020.119715

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