

# CH<sub>3</sub>OH/NH<sub>3</sub>: A Possible Green Energy

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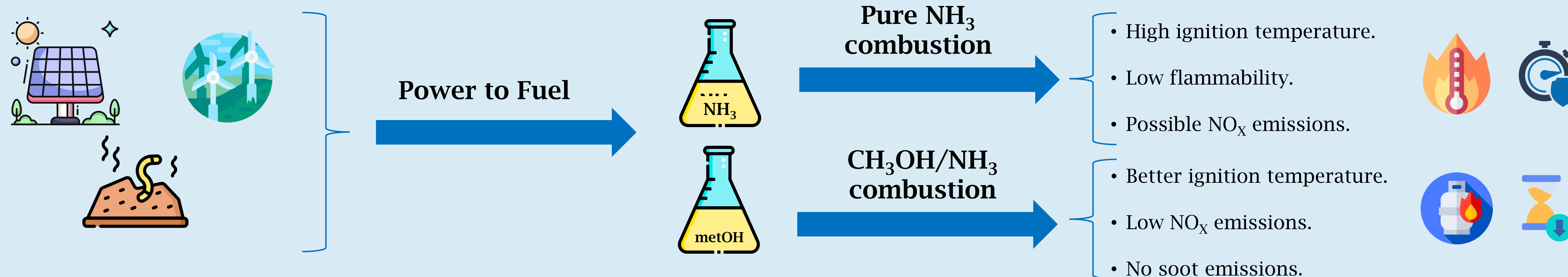
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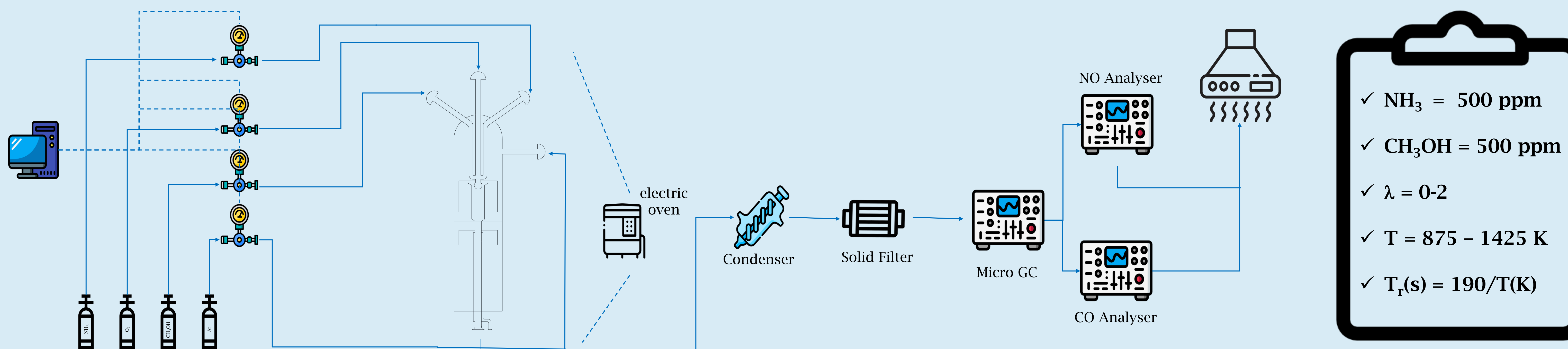
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## Introduction



## Methodology



## Experimental + Simulation

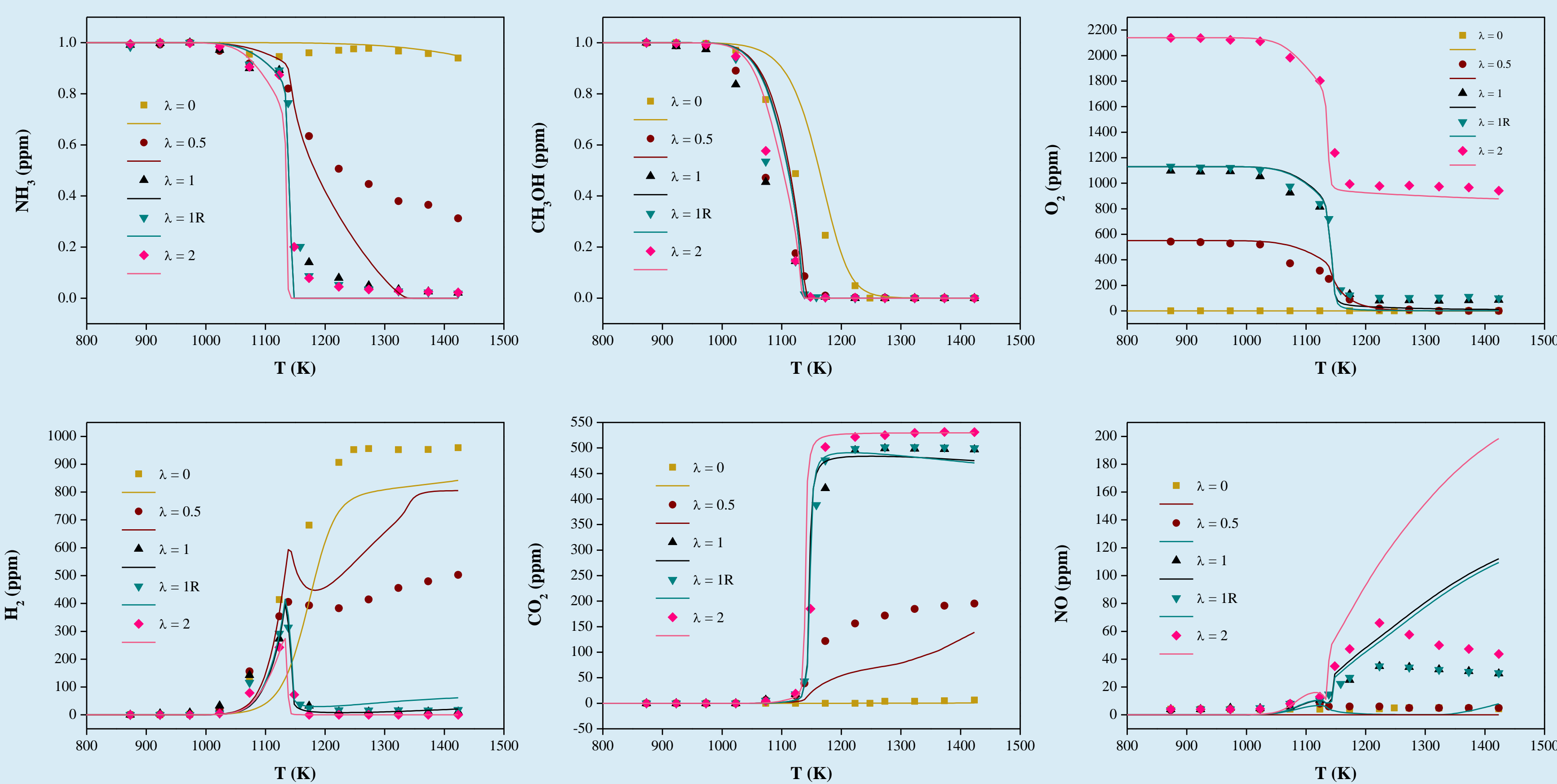


Figure 1. Concentration of different compounds as a function of temperature and excess oxygen ratio.

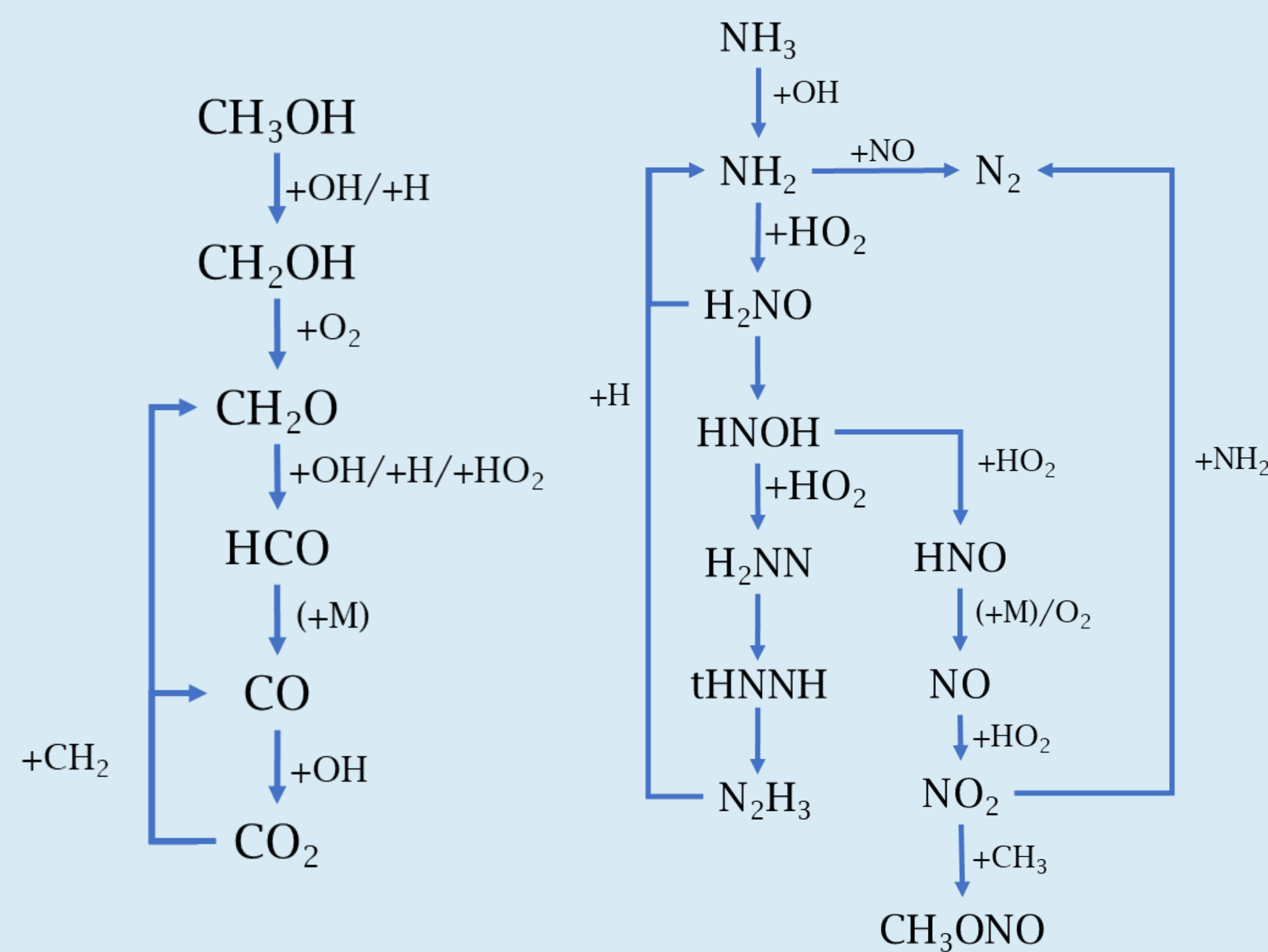


Figure 2. Reaction pathways of NH<sub>3</sub> and CH<sub>3</sub>OH.

## Conclusions



- ✓ Improvement of ammonia ignition.
- ✓ Good prediction by the mechanism.
- ✓ Low NO<sub>x</sub> emissions.
- ✓ Only one major interaction between CH<sub>3</sub>OH and NH<sub>3</sub>.
- ✓ Fuel-lean condition causes oxidation of the compounds at lower temperatures.

## References

- [1] ABIÁN, M., BENÉS, M., de GOÑI, A., MUÑOZ, B., ALZUETA, M.U. Study of the oxidation of ammonia in a flow reactor. Experiments and kinetic modeling simulation. *Fuel*. 2021, 300.
- [2] GLARBORG, P., MILLER, J. A., RUSCIC, B., KLIPPENSTEIN, S. J. Modeling nitrogen chemistry in combustion. *Progress in Energy and Combustion Science* 2018, 67, 31-68.
- [3] MORGAN, E., MANWELL, J., MCGOWAN, J. Wind-powered ammonia fuel production for remote islands: A case study. *Renewable Energy* 2014, 72, 51-61. doi:10.1016/j.renene.2014.06.034

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